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## Total synthesis of bioactive frustulosin and frustulosinol

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Abstract—Stereum hirsutum is a pathogenic fungus involved in 'Esca', a trunk wood disease of grapevine. Among the metabolites isolated from the culture medium of this fungus, frustulosin (2a) shows a high phytotoxicity. A new simple and efficient method for the synthesis of frustulosin (2a) and frustulosinol (2b) was developed. Antibacterial activity and phytotoxicity are reported. 2005 Elsevier Ltd. All rights reserved.

Different fungi infect grapevines and cause severe diseases leading to important economical losses. One of the oldest known is Esca.<sup>[1](#page-2-0)</sup> This disease, progressing worldwide, is associated with a group of fungi that includes Stereum hirsutum.<sup>[2](#page-2-0)</sup> To find an alternative control method against Esca, as efficient as the highly toxic sodium arsenite treatment, now forbidden, the search for the pathogenically active secondary metabolites appears to be necessary. S. hirsutum seems to play an important role in the wood deterioration process[.2](#page-2-0) In a previous work, $3$  we described the isolation and the identification of seven secondary metabolites from the culture medium, and the synthesis of two phytotoxic acetylenic compounds, sterehirsutinal (1a) and sterehirsutinol (1b) (Scheme 1).<sup>4</sup> Frustulosin (2a) and frustulosinol (2b) were also isolated from the same toxic medium. These compounds were previously described as metabolites of Stereum frustulosum and exhibited antimicrobial activity at modest concentrations, particularly against Staphylococ-cus aureus, several Bacilli and Vibrio cholera.<sup>[5](#page-2-0)</sup>

In order to understand the role played by compounds 2a and 2b, large amount of material is required for more



Scheme 1. Phytotoxins extracted from Stereum hirsutum.

specific biological assays. We thus undertook the synthesis of these two compounds 2a and 2b. Two different pathways were already described for the synthesis of frustulosin  $(2a)$ .<sup>[6,7](#page-2-0)</sup> Herein, we report a simpler and more efficient procedure.

Ronald and co-workers described the preparation of 2 iodo-3,6-dihydroxybenzaldehyde in five steps as inter-mediate in the frustulosin (2a) synthesis.<sup>[7](#page-2-0)</sup> Then, the key step was the coupling reaction between this iodo derivative and appropriate alkyne to lead to frustulosin (2a). Thus, we decided to use bromo analogue 4 as key synthon, obtained in one step from commercially available 2,5-dihydroxybenzaldehyde  $(3)$  (Scheme 2).<sup>[8](#page-2-0)</sup> This



Scheme 2. Reagents and conditions: (a)  $Br<sub>2</sub>$ , CHCl<sub>3</sub>, rt, 3 h; (b) TBDMSCl, imidazole, CH<sub>2</sub>Cl<sub>2</sub>, rt, 15 min; (c) 2-methylbut-1-en-3-yne, CuI,  $[PdCl_2(PPh_3)_2]$ , Et<sub>3</sub>N, DMF, 110 °C, 3 h; (d) MOMCl,  $iPr_2EtN$ ,  $CH<sub>2</sub>Cl<sub>2</sub>$ , rt, 1 h.

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bromination reaction afforded exclusively 2-bromo-3,6 dihydroxybenzaldehyde (4) in 97% yield.

Ronald and co-workers achieved the substitution of the iodide by the alkenynyl chain thanks to copper(I) isopropenylacetylide in DMF. A low conversion was observed and a large amount of starting material was recovered. A mixture of frustulosin (2a) (less than 10% yield) and benzofuran derivative 7 was obtained. Using the same procedure, we attempted this transformation with the same poor results. We investigated the crosscoupling reaction between 2-methylbut-1-en-3-yne and the aryl bromide 4 by the Sonogashira reaction<sup>[9](#page-2-0)</sup> in the presence of cuprous iodide, bis(triphenylphosphine) palladium dichloride and triethylamine in DMF. Unfortunately, the results were similar to the cuprate pathway and we obtained mainly the benzofuran derivative 7.

In order to avoid the formation of 7, the hydroxyl group in meta position of the aldehyde had to be protected. TBDMS protecting group was chosen and introduced according to the Hu procedure.[8](#page-2-0) This method allowed a regioselective silylation of the most reactive hydroxyl group of 4, the hydroxyl group in ortho position of bromide. Indeed, the other hydroxyl group is less acidic because of an intramolecular H-bond with aldehyde function that stabilizes the hydroxyl group. Thus, compound 5 was obtained in 97% yield ([Scheme 2\)](#page-0-0).

At this step, the 2-methylbut-1-en-3-ynyle chain had to be introduced into compound 5. No reaction occurred when the Sonogashira cross-coupling was realized at 70  $\degree$ C. But beyond this temperature, the silyl ether was cleaved and the benzofuran derivative 7 formed as the major product [\(Scheme 2\)](#page-0-0). Tentatively, the reaction using the cuprate $6$  was also tested but unsuccessfully.

The MOM group was used to protect hydroxyl groups as in the syntheses of siccayne,<sup>[10](#page-2-0)</sup> eutypine,<sup>[11](#page-2-0)</sup> and sterehirsutinal, $\frac{4}{3}$  $\frac{4}{3}$  $\frac{4}{3}$  that include a Sonogashira reaction for the acetylenic side chain introduction. Thus, we had obtained the protected compound 6 in 75% yield ([Scheme 2\)](#page-0-0). No reaction occurred in Sonogashira conditions between 6 and acetylenic derivative.

The regioselective protection of the most nucleophilic hydroxy group of 4 was achieved with 1.05 equiv of MOMCl affording 8 in 80% yield (Scheme 3). The Sonogashira reaction between bromide 8 and 2-methylbut-1 en-3-yne led to the protected frustulosin 9 in 79% yield.<sup>[12](#page-2-0)</sup> In this case, the presence of MOM group in meta position of the bromine atom is deactivating for the Sonogashira reaction. Finally, frustulosin (2a) was obtained after MOM cleavage using a solution of 10% TFA in CH<sub>2</sub>Cl<sub>2</sub> (v/v) in 60% yield.<sup>[13](#page-2-0)</sup> Biologically interesting frustulosinol (2b) was obtained in 62% yield by chemioselective reduction of frustulosin (2a) with  $NaBH<sub>4</sub>$  in aqueous medium.<sup>14</sup>

Preliminary phytotoxic tests were carried out on grapevine leaf discs. Solutions of frustulosin (2a) and frustulosinol (2b) in 98:2  $H_2O$ /ethanol were prepared in



**Scheme 3.** Reagents and conditions: (a) MOMCl,  $iPr_2E$ tN, CH<sub>2</sub>Cl<sub>2</sub>, rt, 1 h; (b) 2-methylbut-1-en-3-yne, CuI,  $[PdCl_2(PPh_3)_2]$ , Et<sub>3</sub>N, DMF, 60 °C, 3 h; (c) 10:90 TFA/CH<sub>2</sub>Cl<sub>2</sub> (v/v), rt, 5 h; (d) NaBH<sub>4</sub>, 1% (w/v) aq NaOH soln, rt, 2 h.

different concentrations. The same test was realized simultaneously with eutypine as reference, another fungal toxin of grapevine, previously isolated in our laboratory that shows a high phytotoxic activity.[15](#page-2-0)

Necroses already appeared at different levels on leaves before 18 h at 25 °C the concentrations of 500, 250, 100 and 50 mg/L for frustulosin  $(2a)$ . At the concentration of 25 mg/L, necroses were observed only between 18 and 24 h. The appearance of necroses caused by eutypine was less important and no phytotoxicity was observed at the concentration of 25 mg/L. Frustulosinol (2b) was less toxic, necroses were only observed at 250 and 500 mg/L.

Compounds were also tested on grapevine callus (kindly supplied by Roustan, INRA, Toulouse, France). Callus cultures of Vitis vinifera cv. Gamay were grown on a synthetic medium, 12 h daylight for 28 days at 28  $\degree$ C. Calli were weighed before and after incubation with different concentrations of 2a, 2b and eutypine and percent growth was calculated. Frustulosin  $(2a)$  and frustulosinol (2b) inhibit 100% callus growth at 20 mg/L and 50 mg/L, respectively.

In addition, antibacterial tests were realized with frustulosin (2a). Activities were measured on TLC by bioautography on thin-layer plates. Test solutions  $(10 \mu L)$ were applied as small spots on TLC plates (Silicagel G) to give a concentration series of  $0.1-30 \mu g$ /application zone. The organic solvent was evaporated by steam air and plates were eluted by the adequate solvent to obtain different  $R_f$ s. The solvent was evaporated once again and the plates homogeneously sprayed with 10 mL of nutrient agar infected with 1 ml of the nutrient broth containing bacteria  $(10^8 - 10^9 \text{ CFU/ml})$ . Plates were incubated for 2 days at 30 and 37  $\mathrm{^{\circ}C}$  in the dark. The appearance of a blank after spraying the plate with a solution of thiazolyl blue tetrazolium bromide, indicated antibacterial activity. Minimum inhibitory concentrations (MIC) against Bacillus subtilis and Escherichia coli were determined from the lowest tested compound concentration causing recognizable bacterial growth inhibition. Chloramphenicol was used as posi-

<span id="page-2-0"></span>tive control and acetone as negative. Antibacterial activity for frustulosin (2a) was observed at 3 and 10  $\mu$ g, respectively, against B. subtilis and E. coli.

In conclusion, this new synthetic route supplies rapidly a large amount of frustulosin (2a) with a total yield of 37% in only four steps. Frustulosinol (2b) is also obtained in good yield. This procedure can be applied to the synthesis of other phytotoxins belonging to the same family, that is, sterehirsutinal, isolated from S. hirsutum. Biological activities of synthesized compounds have been checked. Phytotoxicity has been exhibited on grapevine and the antibacterial activity has been shown against B. subtilis and E. coli. The efficiency of the method allows the large scale synthesis of such compounds in order to investigate more advanced biological tests.

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- 12. Sonogashira reaction procedure: 2-Hydroxy-5-(methoxymethyloxy)-6-(3-methylbut-3-en-1-ynyl)benzaldehyde 9. A solution of 8 (4.73 g, 18.1 mmol, 1.0 equiv), CuI  $(1.72 \text{ g}, 9.06 \text{ mmol}, 0.5 \text{ equiv}), [PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>] (3.18 \text{ g},$

4.53 mmol, 25 mol %), Et<sub>3</sub>N (10.1 mL, 72.5 mmol, 4.0) equiv) and 2-methylbut-1-en-3-yne (6.9 mL, 72.5 mmol, 4.0 equiv) in 72 mL of DMF was heated at 60 °C under  $\rm N_2$ atmosphere for 3 h. The mixture was then diluted with  $Et<sub>2</sub>O$ , washed twice with 5% aq HCl soln and satd aq NaCl soln, dried over MgSO<sub>4</sub>, and concentrated. Purification by flash chromatography on silica gel (eluant: nhexane/acetone 97:3) provided 3.52 g (79%) of 9 as an orange oil: IR (film) 2924, 1652, 1466, 1272, 1154, 1016; <sup>1</sup>H NMR  $\delta$  (CDCl<sub>3</sub>, 400 MHz): 11.40 (s, 1H, C<sub>6</sub>-OH), 10.42 (s, 1H, CHO), 7.33 (d,  $J = 9.1$  Hz, 1H, C<sub>4</sub>-H), 6.89<br>(d,  $J = 9.1$  Hz, 1H, C<sub>5</sub>-H), 5.48 (qt<sub>app</sub>,  ${}^{2}J = {}^{4}J = 1.0$  Hz,<br>1H, CH<sub>2</sub>=), 5.40 (qt<sub>app</sub>,  ${}^{2}J = {}^{4}J = 1.6$  Hz, 1H, CH<sub>2</sub>=), 5.19 (s, 2H, OCH<sub>2</sub>O), 3.56 (s, 3H, OCH<sub>3</sub>), 2.03 (dd,  ${}^{4}J_{\text{cisoide}} = 1.1 \text{ Hz}, {}^{4}J_{\text{transoide}} = 1.4 \text{ Hz}, {}^{3}H, {}^{3}C$ NMR δ (CDCl<sub>3</sub>, 100 MHz): 197.2, 158.2, 151.6, 127.1, 126.7, 123.8, 119.5, 119.0, 117.6, 102.3, 96.8, 80.1, 56.8, 23.5; MS (ESI<sup>-</sup>)  $m/z = 245.1$  [M-H]<sup>-</sup>.

- 13. Deprotection of compound 9: 2,5-Dihydroxy-6-(3-methylbut-3-en-1-ynyl)benzaldehyde (frustulosin) 2a: To a solution of 9 (1.85 g, 7.51 mmol, 1.0 equiv) in 67.5 mL of  $CH_2Cl_2$  was added 7.5 mL of TFA at 0 °C. The reaction mixture was stirred for 6 h and then TFA was removed in vacuo. Purification by flash chromatography on silica gel (eluant: *n*-hexane/EtOAc 95:5) provided 915 mg  $(60\%)$  of frustulosin 2a as a yellow solid: IR (KBr disc) 3281, 2923, 1641, 1464, 1280, 1149; <sup>1</sup>H NMR δ (CDCl<sub>3</sub>, 400 MHz): 11.25 (s, 1H, C<sub>6</sub>-OH), 10.33 (d, <sup>5</sup>J = 0.5 Hz, 1H, CHO), 7.19 (d,  $J = 9.1$  Hz, 1H,  $C_4$ –H), 6.92 (dd,  $3J = 9.1$  Hz,  $5I = 0.5$  Hz, 1H C, H), 5.58 (e, 1H C, OH), 5.54 (gt,  $J = 0.5$  Hz, 1H, C<sub>5</sub>-H), 5.58 (s, 1H, C<sub>3</sub>-OH), 5.54 (qt<sub>app</sub>,  $^{2}J = {}^{4}J = 1.0$  Hz, 1H, CH<sub>2</sub>=), 5.40 (qt<sub>app</sub>,  $^{2}J = {}^{4}J = 1.6$  Hz, 1H, CH<sub>2</sub>=), 2.06 (dd,  ${}^{4}J_{\text{cisoid}} = 1.1$  Hz,  ${}^{4}J_{\text{transoid}} =$ 1.5 Hz, 3H, CH<sub>3</sub>);  $\delta$  (CDCl<sub>3</sub>, 100 MHz): 196.3, 157.1, 150.6, 125.9, 124.9, 124.8, 120.0, 118.3, 110.7, 104.4, 78.3, 23.6; MS (ESI<sup>-</sup>)  $m/z = 201.5$  [M-H]<sup>-</sup>.
- 14. Reduction of frustulosin (2a): 2-Hydroxymethyl-3-(3 methylbut-3-en-1-ynyl)benzene-1,4-diol (frustulosinol) 2b: To 2a (85 mg, 0.42 mmol, 1.0 equiv) in 1% aq NaOH soln (10 mL) was added dropwise 10% aq NaOH soln (1 mL) of NaBH<sub>4</sub> (21 mg,  $0.55$  mmol, 1.3 equiv). After 2 h, the reaction mixture was diluted with EtOAc and acidified with 5% aq HCl soln. The organic layer was then washed with aq  $NaHCO<sub>3</sub>$  soln, dried over  $MgSO<sub>4</sub>$  and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (eluant: n-hexane/EtOAc 80:20) to give frustulosinol 2 (86 mg,  $62\%$ ) as a yellow solid: IR (KBr disc) 3371, 2937, 1473, 1237, 1169; <sup>1</sup>H NMR  $\delta$  (CDCl<sub>3</sub>, 400 MHz): 7.61 (s, 1H, OH), 6.78 (s, 2H,<br>C<sub>5</sub>-H, C<sub>6</sub>-H), 5.52 (s, 1H, OH), 5.44 (qt<sub>app</sub>, <sup>2</sup> $J = {}^4J =$ <br>0.8 Hz, 1H, CH<sub>2</sub>=), 5.38 (qt<sub>app</sub>, <sup>2</sup> $J = {}^4J = 1.5$  Hz, 1H,<br>CH<sub>2</sub>=), 2.94 (s, 1H, OH), 2.02 (dd, 100 MHz): 150.2, 149.5, 125.9, 124.8, 123.2, 118.6, 114.8, 107.8, 101.7, 79.9, 62.6, 23.4; MS (ESI<sup>-</sup>)  $m/z = 203.6$  $[M-H]$ .
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